

Homework 2 answers

① See Landau-Lifshitz, v. 5, § 39

② a)
$$V(\varepsilon) = \frac{m}{2\pi\hbar^2}$$

b)
$$\langle \varepsilon \rangle = T$$

c)
$$\langle \varepsilon^2 \rangle = \frac{\int_0^\infty e^{-\frac{\varepsilon}{T}} \varepsilon^2 d\varepsilon}{\int_0^\infty e^{-\frac{\varepsilon}{T}} d\varepsilon} = 2T^2$$

③
$$\langle \hat{S}_z \rangle = \frac{2 \sinh\left(\frac{B}{T}\right)}{1 + 2 \cosh\left(\frac{B}{T}\right)}$$

④
$$\langle \varepsilon \rangle = \frac{\hbar\omega}{e^{\frac{\hbar\omega}{T}} - 1} + \frac{\hbar\omega}{2}$$

⑤
$$\langle U \rangle = T$$

Boltzmann distribution: $n = n_0 e^{-\frac{mgz}{T}}$
 Total number of particles: $(n_0 = \frac{P}{T})$

$$N = \int_0^\infty n_0 e^{-\frac{mgz}{T}} S dz = n_0 S \frac{T}{mg} = \frac{PS}{mg}$$

$$C = N \left(\frac{\partial \langle \mathcal{E} \rangle}{\partial T} + \frac{\partial \langle U \rangle}{\partial T} \right) = N \left(\frac{5}{2} + 1 \right) = \frac{7}{2} N$$

assuming diatomic molecules

Note: $C = C_p$ (regardless of any assumptions about molecules in the air)